This article was downloaded by: [University of Haifa Library]

On: 13 August 2012, At: 20:32 Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered

office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gmcl20

Large Two-Photon Absorption Properties of Polyphenyls and Polyfluorenes

Rémi Anémian ^a , Patrice L. Baldeck ^b & Chantal Andraud ^a ^a StÉrÉchimie et Interactions molÉculaires, UMR CNRS, École Normale SupÉrieure de Lyon, n° 5532, Lyon cedex 07, 69364, France

^b Spectrométrie Physique, Université Joseph Fourier, CNRS (UMR 5588), B.P. 87, St Martin d'Hères, 38402, France

Version of record first published: 29 Oct 2010

To cite this article: Rémi Anémian, Patrice L. Baldeck & Chantal Andraud (2002): Large Two-Photon Absorption Properties of Polyphenyls and Polyfluorenes, Molecular Crystals and Liquid Crystals, 374:1, 335-342

To link to this article: http://dx.doi.org/10.1080/713738231

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Large Two-Photon Absorption Properties of Polyphenyls and Polyfluorenes

RÉMI ANÉMIAN^a, PATRICE L. BALDECK^b and CHANTAL ANDRAUD^a

 ^aÉcole Normale Supérieure de Lyon, Stéréochimie et Interactions moléculaires, UMR CNRS n° 5532, 69364 Lyon cedex 07, France and
 ^bSpectrométrie Physique, Université Joseph Fourier, CNRS (UMR 5588), B.P. 87, 38402 St Martin d'Hères, France

The theoretical study of two-photon absorption (TPA) properties of polyphenyls and polyfluorenes (with the number of phenyl rings N = 2 to 6) was related to cooperative effects between monomers. These effects lead to an increase of the TPA cross-section with N without saturation according to a power law.

<u>Keywords</u> Two-photon absorption; polyfluorenes; polyphenyls; <u>CNDO/S</u> calculations.

INTRODUCTION

Two-photon absorption (TPA) properties of organic compounds are being involved in many applications [1]. Several organic molecules optimized for these applications, consisted generally in symmetrically substituted molecules, in which the role of electroactive groups was studied [2-4].

Recently efficient nonlinear absorption properties based on TPA processes were found in unsubstituted dimers, a bistilbene molecule (the Stilbene-3) [5], and the bifluorene [6], and a polyfluorene (60 monomers units) [7].

In this paper, we present theoretical TPA properties, using semiempirical CNDO/S calculations, of two oligomers systems, the polyphenyls I and the polyfluorenes II (Fig. 1), in order to dissect the TPA cross-section on the basis of the two-level model and provide the origin of the enhanced properties of these systems. Preliminary results concerning this study were reported for polyphenyls [8].

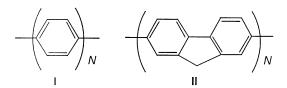


FIGURE 1 Molecular structures of oligomers I_N and II_N

CALCULATIONS METHOD

The molecule geometry was optimized by using Sybyl, i.e. Tripos force field and AM1 from the MOPAC package [9]; the selected geometry corresponds to the lowest energy. For each molecule, the electronic state characteristics, i.e. singlet state energies, and dipole moments, were obtained by a configuration interaction (CI) procedure based on the CNDO/S method using the QCPE program #333. The CI calculations included 100 singly (SCI), and 100 doubly (DCI) excited configurations.

The TPA cross-section σ_{TPA} is proportional to the imaginary part of the average second hyperpolarizability:

$$\sigma_{TTA} = \frac{3}{10} \hbar \frac{\omega^2}{n^2 c^2 \varepsilon_0} f^4 \operatorname{Im} \langle \gamma(-\omega; \omega, -\omega, \omega) \rangle$$

$$\langle \gamma \rangle = \frac{1}{5} \left(\gamma_{xxxx} + \gamma_{yyyy} + \gamma_{zzzz} + \gamma_{xxyy} + \gamma_{yyxx} + \gamma_{yyzz} + \gamma_{zzyy} + \gamma_{xxzz} + \gamma_{zzxx} \right)$$
(2)

The tensor components $\gamma_{\rm jkl}$ were computed using Orr and Ward's equation based on the time-dependent perturbation theory [10]. TPA spectra were evaluated by using the parameters of the 19 lowest excited states to obtain a good convergence of σ_{TPA} values (Fig. 3). The damping factors Γ were taken to be 1000 cm⁻¹ for all states, and the refractive index was assumed to be 1.

Results were interpreted on the basis of the three-level model (Eq.(3)and Fig. 2):

$$\gamma \propto \frac{\mu_{01}^2 \mu_{12}^2}{(E_{01} - \frac{E_{02}}{2})^2} \tag{3}$$

where μ_{01} , and μ_{12} , are the transition moment between the ground state S_0 and the lowest charge transfer excited-state S_1 , and the transition moment between S_1 and the lowest two-photon excited state S_2 , respectively. E_{01} and E_{02} are the energy of S_1 and S_2 , respectively. [3]

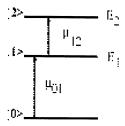


FIGURE 2 Three-level model scheme

As an illustration of the validity of this model for oligomers systems, variations of σ_{TPA} with the number of excited states was reported on Fig. 3 for the bifluorene $\mathbf{I_2}$ and tetraphenyl $\mathbf{II_2}$ molecules. A rapid convergence of σ_{TPA} is obtained for both molecules. Negative values corresponding to one-photon resonance saturation effects were obtained up to 5 excited states, while the maximum σ_{TPA} value was found for 7 states; the subsequent decrease was ascribed to negative contribution of μ_{0i} μ_{ij} μ_{jk} μ_{ko} terms.

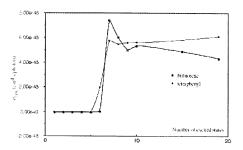


FIGURE 3 TPA cross-section as a function of the number of excited states included in calculations

RESULTS

The spectral distribution of σ_{TPA} for oligomers I and II is presented on Fig. 4 for N=2 and 6. The theoretical spectrum consists mainly in a band centered at an energy of $\frac{E_{02}}{2}$ (Eq(3)). A weak wavelength shift is obtained with N for the spectral distribution of σ_{TPA} in the case of the polyfluorenes II (see also Fig.6).

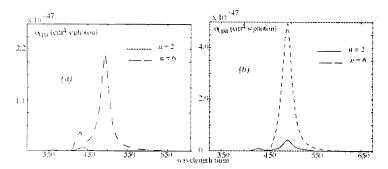


FIGURE 4 TPA spectrum for I (a) and II (b).

Values of σ_{TPA} are reported on Table 1 as a function of N. When comparing values of σ_{TPA} for a same number of monomers N (a benzene ring for \mathbf{I} and a fluorene for \mathbf{II}), \mathbf{II} leads to higher values σ_{TPA} .

N	I	II	
2	80	430	
3	330	1502	
4	610	2289	
5	1260	3290	
6	2090	4940	

TABLE 1 TPA cross-sections (10^{-50} cm⁴ s/photon) of polyphenyls and polyfluorenes as a function of the monomers number N.

Variations of σ_{TPA} are presented on Fig. 5. A In order to compare TPA properties of **I** and **II**, these variations were plotted versus N' according to the system (4).

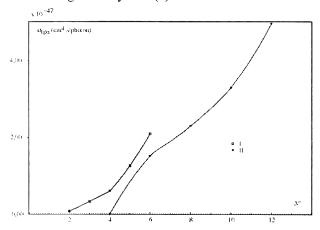


FIGURE 5 σ_{TPA} versus the number of moieties for I and II

$$N' = N$$
$$N' = 2N$$

A continuous increase of σ_{TPA} was obtained for N = 1 to 6 for I and II. However, when comparing values for a same number of benzene rings N', polyphenyls I give rise higher values σ_{TPA} (Fig. 5). These large variations of σ_{TPA} could be quantified following the power law (5):

$$P = KN^{\alpha} \tag{5}$$

These variations are in good agreement with recent theoretical results obtained for polyene oligomers, for which a strong correlation between TPA properties and the length of theses systems was established [11]. Values of the exponent α obtained for **I** and **II** for σ_{TPA} are reported on Table 2. These values of α are in good agreement with found for other oligomers ($\alpha = 5$ -6[7c]).

\overline{P}	σ_{TPA}	μ01	μ02	$(E_{01} - \frac{E_{02}}{2})^{-1}$	$\sigma_{TPA}^{3-level}$
II	2.8	0.67 0.54	0.64 0.42	0.47 0.25	3.6 2.4

TABLE 2 Exponents α involved in Eq. (5) for parameters P describing the relationship (3) in oligomers **I** and **II**

DISCUSSION

In the following we will consider oligomers **I** and **II** as constituted of N monomers in excitonic coupling similarly to J-aggregates in order to interpret the above enhancement of σ_{TPA} of these oligomers. This assumption requires a dependence of the dipole moments μ_{ij}^N of the oligomer with that of the monomer μ_{ij}^1 following the relationship (6) [12,13]:

$$\mu_{ij}^N = \sqrt{N} \mu_{ij}^1 \quad (6)$$

Values of μ_{01} and μ_{12} are reported on Fig. 6 for systems **I** and **II** with that of E_{01} and E_{02} . Variations with N of these parameters follows a power law of type (5), for which the values of the exponent α are reported on Table 2. These values, which are very close for μ_{01} and μ_{12} from the value of 0.5 proposed in the excitonic model (see Eq. (6)), valid our assumption of coupling interactions in oligomers **I** and **II**.

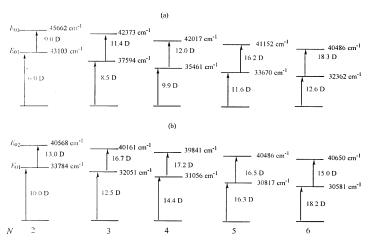


FIGURE 6 Energy diagram with μ_{01} , μ_{12} values for systems **I** (a) and **II** (b).

We note on Fig. 6 that polyfluorenes II presents energies E_{01} and E_{02} shifted towards the red range with respect to those of a polyphenyls I. However, for systems II weaker variations with N of these energies are predicted; furthermore, a decrease with N of μ_{12} is calculated. These observations is in good agreement with preliminary experimental data in fluorenes [14] and led to weaker α exponent values for μ_{01} , μ_{12} and $(E_{01} - \frac{E_{02}}{2})^{-1}$ in the case of systems II (Table 2).

The above coupling model allows to interpret readily large variations of σ_{TPA} with N; when replacing all parameters (μ_{01} , μ_{12} and ($E_{01} - \frac{E_{02}}{2}$)⁻¹) involved in the relationship of the three-level model (Eq. 3) by their power law expression, we obtained an α exponent value for $\sigma_{TPA}^{3, level}$ similar to that obtained directly from variations of σ_{TPA} with N (Table 2).

CONCLUSION

This theoretical study shows that the two-photon absorption cross-section of polyphenyls and polyfluorenes increases with the number of monomer units. No saturation is predicted at least up to n=6. This TPA enhancement follows a power law that results from cooperative effects between monomers on excited state characteristics.

REFERENCES

[1] B.A. Reihardt Photonics Science News 4, 21 (1999).

[2] (a) J.E. Ehrlich, X.L. Wu, I.-Y.S. Lee, Z.-Y. Hu, H. Röckel, S.R. Marder, J.W. Perry Opt. Lett. 22, 1843 (1997); (b) J.W. Perry, S. Barlow, J.E. Ehrlich, A.A. Heikal, Z.-Y. Hu, I.-Y.S. Lee, K. Mansour, S.R. Marder, H. Röckel, M. Rumi, S. Thayumanavan, X.-L. Wu Nonlinear Optics 21, 225 (1999); (c) S. Delysse, P. Filloux, V. Dumarcher, C. Fiorini, J.-M. Nunzi Optical Materials 9, 347 (1998); (d) P.-A. Chollet, V. Dumarcher, J.-M. Nunzi, P. Feneyrou, P. Baldeck Nonlinear Optics 21, 299 (1999); (e) C. Nguefack, T. Zabulon, R. Anémian, C. Andraud, A. Collet, S. Topçu, P.L. Baldeck Nonlinear Optics 21, 309 (1999).

[3] M. Albota, D. Beljonne, J.L. Brédas, J.E. Ehrlich, J.-Y. Fu, A.A. Heikal. S.E. Hess, T. Kogej, M.D. levin, S.R. Marder, D. McCord-Maughon, J.W. Perry, H. Röckel, M. Rumi, G. Subramaniam, W.W.

Webb. X.-L. Wu, C. Xu Science **281**, 1653 (1998).

[4] C. Andraud, R. Anémian, A. Collet, J.-M. Nunzi, Y. Morel, P.L. Baldeck J. Opt. A: Pure Appl. Opt. 2, 284 (2000); B. Paci, J.-M. Nunzi. R. Anémian, C. Andraud, A. Collet, Y. Morel, P.-L. Baldeck J. Opt. A: Pure Appl. Opt. 2, 268 (2000)

Opt. A: Pure Appl. Opt. 2, 268 (2000).

[5] P.-A. Chollet, V. Dumarcher, J.-M. Nunzi, P. Feneyrou, and P.L. Baldeck Nonlinear Optics, 21, 299 (1999); P.-A. Chollet, A. Sornin, B. Paci, V. Hully, J.-M. Nunzi, Y. Morel, and P.L. Baldeck, in Mat. Res. Soc. Symp. Proc., Thin Films for Optical Waveguide Devices and Materials for Optical Limiting, edited by K. Nashimoto, B.W. Wessels, J. Shmulovich, A.K-Y. Jen, K. Lewis, R. Pachter, R. Sutherland, J. Perry, (MRS Boston 1999), Vol. 597, in press.

[6] Y. Morel, A. Irimia, P. Najechalski, Y. Kervella, O. Stephan, P.L. Baldeck J. Chem. Phys. in press; P. Najechalski, Y. Morel, O. Stephan,

P.L. Baldeck Chem. Phys. Lett. in press.

[7] P. Najechalski, Y. Morel, O. Štéphan, P. L. Baldeck Chem. Phys. Lett. in press.

[8] C. Andraud, P.L. Baldeck Nonlinear Optics (in press).

[9] SYBYL 6.5, from Tripos, 1699 S. Hanley Road, St. Louis, MO 63144-2913. (Tripos force field ans AM1 calculation included in MOPAC package).

[10] B.J. Orr, J.F. Ward Mol. Phys. **20**, 513 (1971).

[11] (a) A.A. Said, C. Wamsmey, D.J. Hagan, E.W. Van Stryland, B.A. Reinhardt, P. Rodere, A. G. Dillard Chem. Phys. Lett. **228**, 646 (1994); (b) D. Beljonne, J.L. Brédas J. Opt. Soc. Am. B **11**, 1380 (1994); (c) P. Norman, Y. Luo, H. Agren Chem. Phys. Lett. **296**, 8 (1998); (d) P. Norman, Y. Luo, H. Agren J. Chem. Phys. **111**, 1 (1999).

[12] E.G. McRae, M. Kasha J. Chem. Phys. 28, 721 (1958).

[13] A.S. Davidov Zhur. Eksptl. I Teort. Fiz. 18, 515 (1948).

[14] Results to be published.